

# Regulation of cellular $\text{Mg}^{2+}$ by *Saccharomyces cerevisiae*

Troy Beeler<sup>\*</sup>, Kerry Bruce, Teresa Dunn

Department of Biochemistry, Uniformed Services University of the Health Sciences, 4301 Jones Bridge Road, Bethesda, MD 20814, USA

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## Abstract

Regulation of cellular  $\text{Mg}^{2+}$  by *S. cerevisiae* was investigated. The minimal concentration of  $\text{Mg}^{2+}$  that results in optimal growth of *S. cerevisiae* is about 30  $\mu\text{M}$  and a half-maximum growth rate is attained at about 5  $\mu\text{M}$   $\text{Mg}^{2+}$ . Since the plasma membrane has an electrical potential greater than 100 mV, passive equilibration of  $\text{Mg}^{2+}$  across the plasma membrane would provide sufficient cytosolic  $\text{Mg}^{2+}$  (0.1–1 mM). The total cellular  $\text{Mg}^{2+}$  of cells grown in synthetic medium containing 1 mM  $\text{Mg}^{2+}$  is about 400 nmol/mg protein, most of which is bound to polyphosphate, nucleic acids, and ATP. Total cellular  $\text{Mg}^{2+}$  decreases to about 80 nmol/mg protein as the  $\text{Mg}^{2+}$  in synthetic growth medium is reduced to 0.02 mM, but remains relatively constant in growth medium containing 1 to 100 mM  $\text{Mg}^{2+}$ . Cells shifted into  $\text{Mg}^{2+}$ -free medium continue to grow by utilizing the vacuolar  $\text{Mg}^{2+}$  stores.  $\text{Mg}^{2+}$ -starved cells replenish vacuolar  $\text{Mg}^{2+}$  stores with a halftime of 30 min. following the addition of 1 mM  $\text{Mg}^{2+}$  to the growth medium. The data indicate that cytosolic  $\text{Mg}^{2+}$  is maintained by the regulation of  $\text{Mg}^{2+}$  fluxes across both the vacuolar and plasma membranes.

**Keywords:** Magnesium; Calcium; Polyphosphate; Vacuole; Dichlorophosphonazo III; (*S. cerevisiae*)

## 1. Introduction

Magnesium is an abundant physiological divalent cation of widespread importance in metabolic processes. Eukaryotic cells maintain the cytosolic  $\text{Mg}^{2+}$  concentration in the range of 0.1 to 1 mM by mechanisms yet to be identified [1]. The genetically tractable model eukaryote *S. cerevisiae* may be useful in identifying transport and regulatory proteins required for  $\text{Mg}^{2+}$  homeostasis. As a first step in establishing a screen to identify and characterize mutants with defects in  $\text{Mg}^{2+}$  homeostasis, an assay for cellular  $\text{Mg}^{2+}$  levels [2] was adapted to *S. cerevisiae* and the relationship between the  $\text{Mg}^{2+}$  concentration in the

growth medium and the cellular  $\text{Mg}^{2+}$  content was investigated.

## 2. Materials and methods

### 2.1. Media and strains

CuH3 (*Mata* *ura3-52 his4-619*) and DBY 947 (*Mata* *ura3-52 ade2-101*) *S. cerevisiae* strains in the S288C background were used. Cells were grown in synthetic Wickerham's minimal media (SD) [3]; where indicated,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$  or phosphate was omitted or added to the stated concentration. For SD plates, electrophoresis grade agarose (1.0%) was substituted for Noble agar.

<sup>\*</sup> Corresponding author. Fax: +1 301 2953512.

## 2.2. Measurement of $Mg^{2+}$

$Mg^{2+}$  was measured spectrophotometrically using dichlorophosphonazo III [2]. Cells were grown in a modified SD media without  $Ca^{2+}$ , harvested, washed by centrifugation at 4°C with 0.1 M KCl, 10 mM Pipes (pH 7.0), 1 mM EDTA to remove extracellular divalent cations and then with 0.1 M KCl, 10 mM Pipes (pH 7.0) to remove EDTA. All solutions used in the assay were passed through Chelex 100 columns to remove divalent cations. Cells were suspended to an  $A_{600}$  of 0.5 in 0.1 M KCl, 10 mM Pipes (pH 7.0), 10  $\mu$ M TPEN (tetrakis(2-pyridylmethyl)ethylenediamine), and 20  $\mu$ M dichlorophosphonazo III. TPEN was added to chelate  $Zn^{2+}$ ,  $Cu^{2+}$  or  $Fe^{3+}$  [4]. The difference absorbance of the dichlorophosphonazo III- $Mg^{2+}$  complex was measured at 672 nm using 600 nm as a reference wavelength using a SLM-Aminco DW2c dual wavelength spectrophotometer. After obtaining a baseline, digitonin (0.1 mg/ml) was added to permeabilize cellular membranes, and the released  $Mg^{2+}$  was determined by measuring the change in the difference spectrum. EDTA and  $Mg^{2+}$  were used to calibrate the response of the dye to  $Mg^{2+}$ . The change in the absorbance spectrum following digitonin addition was the same as that obtained by addition of  $Mg^{2+}$ , and different from that obtained by addition of  $Zn^{2+}$ ,  $Cu^{2+}$  or  $Fe^{3+}$ . In parallel experiments using Arsenazo III to measure  $Ca^{2+}$  as previously described [5], there was no release of  $Ca^{2+}$  from the cell upon digitonin addition.

## 2.3. Measurement of $Ca^{2+}$ accumulation by permeabilized cells

Vacuolar  $Ca^{2+}$  accumulation was measured using osmotically shocked partially regenerated spheroplasts as previously described [5]. Partially regenerated spheroplasts were resuspended in 0.1 M KCl, 10 mM Hepes/Pipes/Mes (pH 7.0), 0.6 M sorbitol to a protein concentration of 1.5 mg/ml. The cells were diluted 50-fold into 0.1 M KCl, 10 mM Hepes/Pipes/Mes (pH 7.0), 50  $\mu$ M  $CaSO_4$ , 50  $\mu$ M Arsenazo III, 1 mM Na ATP and the indicated  $Mg^{2+}$  concentration at 27°C. This osmotic shock resulted in permeabilization of the cell plasma membrane.  $Ca^{2+}$  uptake was monitored spectrophotometrically by

measuring the difference absorbance of the  $Ca^{2+}$ -Arsenazo III complex at 660 nm using 685 as reference wavelength. The baseline absorbance was established by addition of 5  $\mu$ M A23187, a  $Ca^{2+}$  ionophore that releases sequestered  $Ca^{2+}$ . The response of Arsenazo III to  $Ca^{2+}$  was calibrated by multiple additions of 2.5  $\mu$ M  $Ca^{2+}$ .

## 2.4. Measurement of $Ca^{2+}$ accumulation by *S. cerevisiae* cells

$Ca^{2+}$  accumulation by whole cells was measured as previously described [5]. Cells were grown to an  $A_{600}$  of 1.0 in YPD medium and diluted into YPD medium + 50 mM  $CaCl_2$  with varying concentrations of  $Mg^{2+}$ . After a 2-h incubation at 26°C, cells were washed by centrifugation (4°C) and resuspended to a density of  $10^7$  cells/ml in 0.1 M KCl, 10 mM Hepes/Pipes/Mes (pH 7.0), 50  $\mu$ M  $CaSO_4$ , and 50  $\mu$ M Arsenazo III.  $Ca^{2+}$  release from the cell was initiated by adding 1 mg/ml of the detergent digitonin which permeabilizes cell membranes.  $Ca^{2+}$  was measured spectrophotometrically by monitoring the difference absorbance of the Arsenazo III- $Ca^{2+}$  complex at 660–685 nm.

# 3. Results

## 3.1. $Mg^{2+}$ -requirement for growth

The effect of the  $Mg^{2+}$  concentration on the growth rate of *S. cerevisiae* cells was measured in  $Ca^{2+}$ -free synthetic dextrose (SD) media (Fig. 1A). The half-maximum growth rate is observed at about 5  $\mu$ M  $Mg^{2+}$ . At 30  $\mu$ M  $Mg^{2+}$ , the growth rate is about 90% that observed at 2000  $\mu$ M. The cytoplasmic free  $Mg^{2+}$  concentration in eukaryotic cells is expected to be 0.1–1 mM [1]; therefore a 20–200-fold  $Mg^{2+}$  gradient across the plasma membrane is formed when the external  $Mg^{2+}$  is 5  $\mu$ M. Proton transport by the plasma membrane  $H^+$ -ATPase is likely to polarize the membrane to greater than  $-100$  mV [6] which has the potential to form a 10 000-fold  $Mg^{2+}$  gradient ( $[Mg^{2+}]_{in}/[Mg^{2+}]_{out} = e^{-100 \text{ mV } nF/RT}$ ). Passive equilibration of  $Mg^{2+}$  across the polarized plasma membrane would be sufficient to provide the necessary cytosolic  $Mg^{2+}$  concentration.

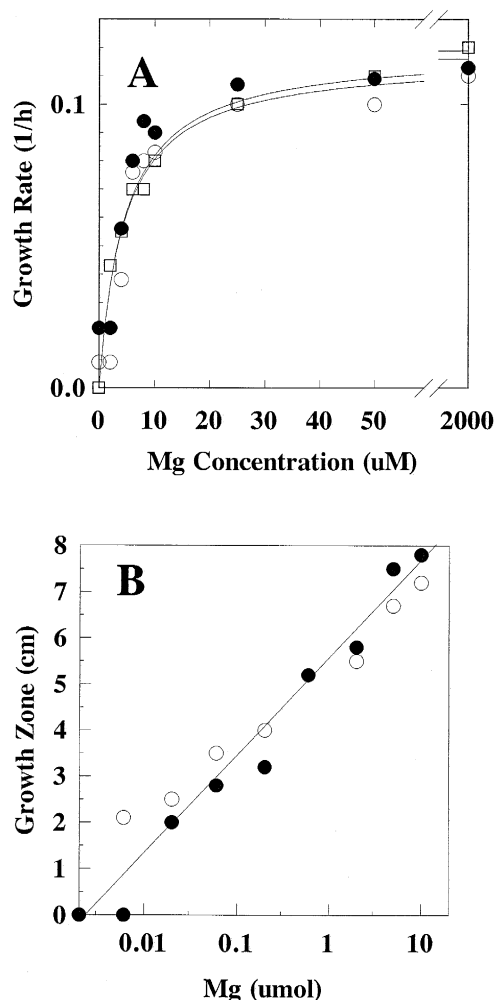


Fig. 1. Effect of the  $\text{Mg}^{2+}$  concentration on the growth rate of *S. cerevisiae* cells. (A) The effect of  $\text{Mg}^{2+}$  concentration on the growth rates of CuH3 (●) and 974 (○, □) cells was determined by monitoring the increase in the  $A_{600}$  over a 10-h period starting with an  $A_{600}$  of 0.1. A modified SD media was used; the  $\text{MgSO}_4$  concentration was varied as indicated and the  $\text{Ca}^{2+}$  concentration was 15  $\mu\text{M}$  (○, ●) or 20 mM (□). Cells were incubated in the media 16 h prior to measuring the growth rate. The growth rate was calculated from the slope of the line formed when the  $\ln A_{600}$  was plotted against time. (B) The effect of  $\text{Mg}^{2+}$  on the growth of cells on SD agarose plates was determined. Cells were grown in YPD and washed with  $\text{Mg}^{2+}$ -free SD media. The cells were diluted to a concentration of  $10^7$  cells/ml and 0.3 ml were spread onto  $\text{Mg}^{2+}$ -free SD agarose plates containing 15  $\mu\text{M}$  or 20 mM  $\text{Ca}^{2+}$ . Paper discs containing 20  $\mu\text{l}$  of a solution containing the indicated amount of  $\text{Mg}^{2+}$  were placed on the plates. The plates were incubated for 3 days at 26°C. The diameter of the zone of growth surrounding the disc was measured (cm).

Addition of 20 mM  $\text{Ca}^{2+}$ , the second most abundant cellular divalent cation, does not significantly alter the dependency of growth on the  $\text{Mg}^{2+}$  concentration in liquid medium (Fig. 1A) or on agarose plates (Fig. 1B) indicating that the  $\text{Mg}^{2+}$  transport systems that mediate cellular  $\text{Mg}^{2+}$  accumulation are not significantly inhibited by  $\text{Ca}^{2+}$ .

### 3.2. Influence of extracellular $[\text{Mg}^{2+}]$ on intracellular $[\text{Mg}^{2+}]$

The relationship between the growth medium  $\text{Mg}^{2+}$  concentration and the cellular  $\text{Mg}^{2+}$  content was determined (Fig. 2). Cells were grown in  $\text{Ca}^{2+}$ -free SD media containing varying amounts of  $\text{Mg}^{2+}$ , washed by centrifugation to remove extracellular  $\text{Mg}^{2+}$ , and placed in Chelex 100-treated KCl solution containing the  $\text{Mg}^{2+}$ -indicator dye, dichlorophosphonazo III.  $\text{Mg}^{2+}$  was released from the cells by permeabilization of the cellular membranes with the detergent digitonin and measured spectrophotometrically by monitoring the difference absorbance of the dichlorophosphonazo III- $\text{Mg}^{2+}$  complex (Fig. 2A). Since  $\text{Ca}^{2+}$  also binds to dichlorophosphonazo III [2],  $\text{Ca}^{2+}$  was omitted from the growth medium. *S. cerevisiae* show normal growth in medium with only trace amounts of  $\text{Ca}^{2+}$ . Under these growth conditions,  $\text{Ca}^{2+}$  is not released from the cells upon permeabilization with digitonin [5]. The chelator tetrakis(2-pyridylmethyl)ethylenediamine (TPEN) was included in the assay solution to prevent heavy metals from interfering with the  $\text{Mg}^{2+}$  determination [4]. Cells grown in media containing 0.3 to 100 mM  $\text{Mg}^{2+}$  accumulate about 400 nmol/mg  $\text{Mg}^{2+}$  (Fig. 2B). This value is similar to that reported by others using atomic absorption to determine cellular  $\text{Mg}^{2+}$  [7–9]. As the growth medium  $\text{Mg}^{2+}$  concentration is reduced from 0.3 mM to 0.03 mM, the cellular  $\text{Mg}^{2+}$  content decreases from about 400 nmol/mg protein to 90 nmol/mg although the growth rate does not change.

### 3.3. The role of the vacuole in $\text{Mg}^{2+}$ storage

Yeast vacuoles are believed to be a major storage site for divalent cations in the cell [11]. The capacity of the vacuole to accumulate divalent cations is greatly increased by its high content of the polyanion

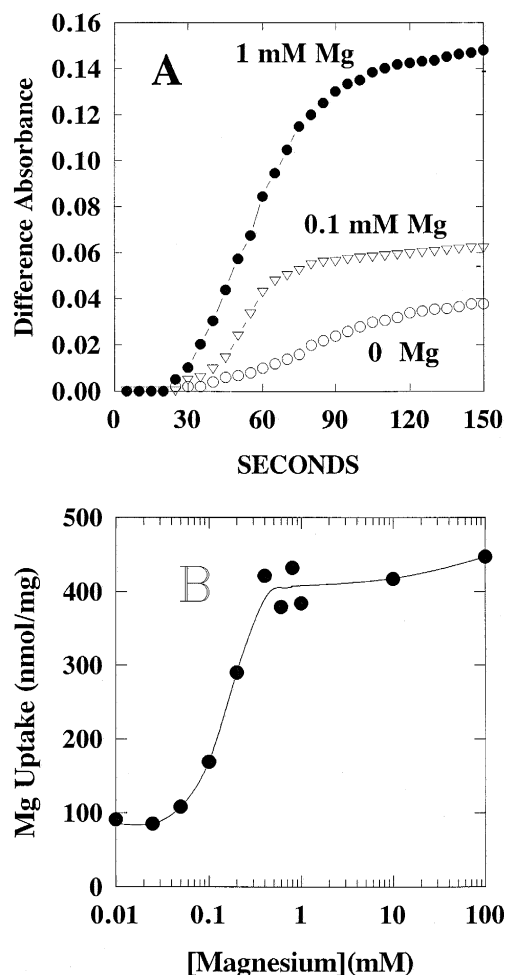


Fig. 2. Effect of the  $Mg^{2+}$  concentration on the level of cellular  $Mg^{2+}$ . (A) Kinetics of digitonin-induced release of cellular  $Mg^{2+}$ . Cells were grown in SD media and then transferred to  $Ca^{2+}$ -free SD medium containing the indicated  $Mg^{2+}$  concentration and incubated for 16 h at 26°C to give a final  $A_{600}$  of 0.2–1.0. The cells were washed by centrifugation three times with 0.1 M KCl, 10 mM Pipes (pH 7.0) and resuspended to  $A_{600}$  of 0.5 in KCl buffer containing 20  $\mu$ M dichlorophosphonazo III and 10  $\mu$ M TPEN. At  $t_0$ , 0.1 mg/ml digitonin was added to permeabilize the organellar and plasma membranes and the difference absorbance of the dichlorophosphonazo III- $Mg^{2+}$  complex (672 nm/600 nm) was monitored to measure the release of cellular  $Mg^{2+}$ . (B) Cells were grown for 16 h to an  $A_{600}$  of 0.1 to 1.0 in  $Ca^{2+}$ -free SD media containing varying amounts of  $Mg^{2+}$ . The cells were then washed with cold KCl buffer, and resuspended in KCl buffer containing 20  $\mu$ M dichlorophosphonazo III and 20  $\mu$ M TPEN. Digitonin (0.1 mg/ml) was added to permeabilize the plasma and organellar membranes to release cellular  $Mg^{2+}$ . The difference absorbance of the  $Mg^{2+}$ -dichlorophosphonazo III complex (672 nm/600 nm) was measured.

polyphosphate (about 1  $\mu$ mol phosphate residue/mg protein) [5]. Cellular  $Mg^{2+}$  can be reduced by 75% without affecting the growth rate suggesting that 75% of the cellular  $Mg^{2+}$  is storage  $Mg^{2+}$ . The vacuole is the likely site for  $Mg^{2+}$  storage.  $Mg^{2+}$  binding to synthetic polyphosphate was measured spectrophotometrically (Fig. 3A). Sodium polyphosphate (80  $\mu$ g/ml, average chain length of 31 residues from Sigma) was titrated with  $Mg^{2+}$  in 0.1 M KCl, 10 mM Pipes buffer (pH 6.5), and 20  $\mu$ M dichlorophosphonazo III. Polyphosphate binds 0.12 mol  $Mg^{2+}$ /mol polyphosphate residue with relatively high affinity ( $K_d$  about 4  $\mu$ M) (Fig. 3B) and more than 0.2 mol  $Mg^{2+}$ /mol polyphosphate with lesser affinity. Since the polyphosphate content of vacuoles is about 0.1 mg polyphosphate/mg protein [5], 120 nmol  $Mg^{2+}$ /mg protein could be complexed with polyphosphate in the vacuole with relatively high affinity while more than 200 nmol  $Mg^{2+}$ /mg could be complexed at lower affinity. The free  $Mg^{2+}$  concentration in the vacuole is not known; however, the vacuolar membrane potential, generated by electrogenic  $H^+$  transport by the vacuolar  $H^+$ -ATPase, is estimated to be 75 mV (inside-positive) [12]. Therefore, if  $Mg^{2+}$  is passively equilibrated across the vacuole membrane and the cytosolic  $Mg^{2+}$  concentration is 1 mM, the vacuolar  $Mg^{2+}$  concentration would be only 3  $\mu$ M. At this free  $Mg^{2+}$ , about 50 nmol  $Mg^{2+}$ /mg cell protein would be bound. However if  $Mg^{2+}$  is transported into the vacuole in exchange for one proton, the luminal  $Mg^{2+}$  concentration would be the same as the cytosolic  $Mg^{2+}$  concentration and the polyphosphate  $Mg^{2+}$  binding sites would be nearly saturated (300–400 nmol  $Mg^{2+}$ /mg protein).

### 3.4. The rate of $Mg^{2+}$ -depletion of cells in $Mg^{2+}$ free solution

When exponentially growing cells are shifted from  $Mg^{2+}$ -containing to  $Mg^{2+}$ -free medium, they continue to grow for about 9 h (Fig. 4A) and then cells begin to die. During this time the  $Mg^{2+}$  content of the cells decreases 4.8-fold from 390 nmol/mg to 80 nmol/mg (Fig. 4B) while the number of viable cells increases 3.5 fold indicating that stored  $Mg^{2+}$  can be utilized to support cell growth. However, growth ceases once cellular  $Mg^{2+}$  is lowered to about 80

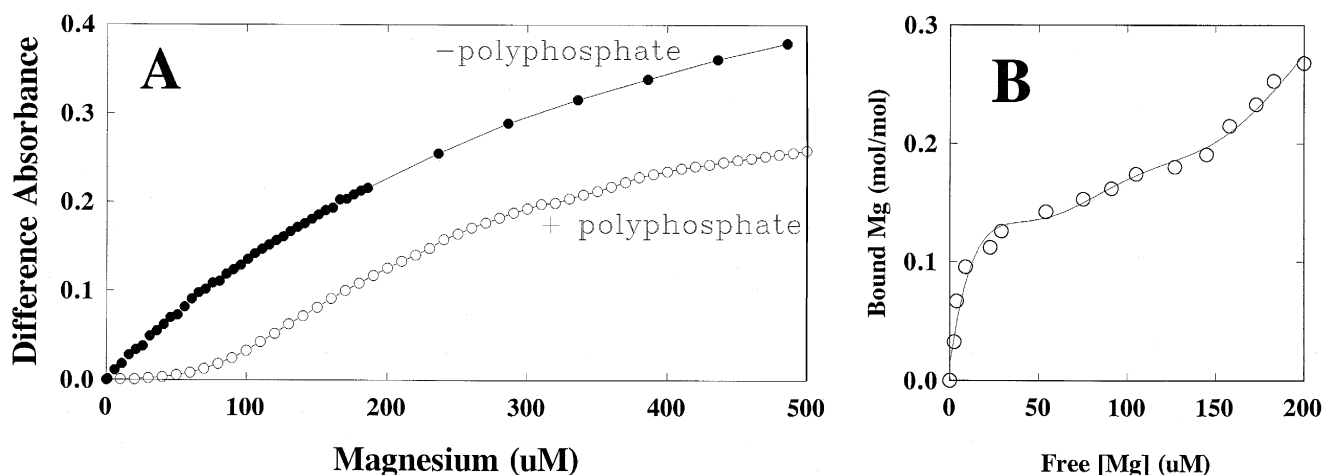


Fig. 3. Mg<sup>2+</sup> binding to synthetic polyphosphate. (A) The Mg<sup>2+</sup>-indicator dichlorophosphonazo III was titrated with Mg<sup>2+</sup> in the presence (○) and absence (●) of 80 μg/ml polyphosphate. The assay solution contained 0.1 M KCl, 10 mM Mes (pH 6.5) and 20 μM dichlorophosphonazo III. The difference absorbance of the Mg<sup>2+</sup>-dichlorophosphonazo III complex (672 nm/600 nm) was measured and plotted against the total Mg<sup>2+</sup> concentration. (B) The polyphosphate-bound Mg<sup>2+</sup> (Total Mg<sup>2+</sup> minus dichlorophosphonazo III-bound and free Mg<sup>2+</sup>) was plotted against free Mg<sup>2+</sup>.

nmol Mg<sup>2+</sup>/mg protein and between 9 and 50 h in Mg<sup>2+</sup>-free YPD, the number of viable cells decreases 3.5-fold.

Cells grown in 20 μM Mg<sup>2+</sup> have only 90 nmol Mg<sup>2+</sup>/mg and stop growth immediately upon transfer to Mg<sup>2+</sup>-free media (Fig. 5) indicating that the

extra 300 nmol Mg<sup>2+</sup>/mg protein in cells grown in 1 mM Mg<sup>2+</sup> is utilized to permit continuous growth in low Mg<sup>2+</sup> medium. Cells grown in media containing only 50 μM phosphate do not accumulate polyphosphate and also stop growth immediately upon transfer to Mg<sup>2+</sup>-free, high-phosphate medium. Cells grown

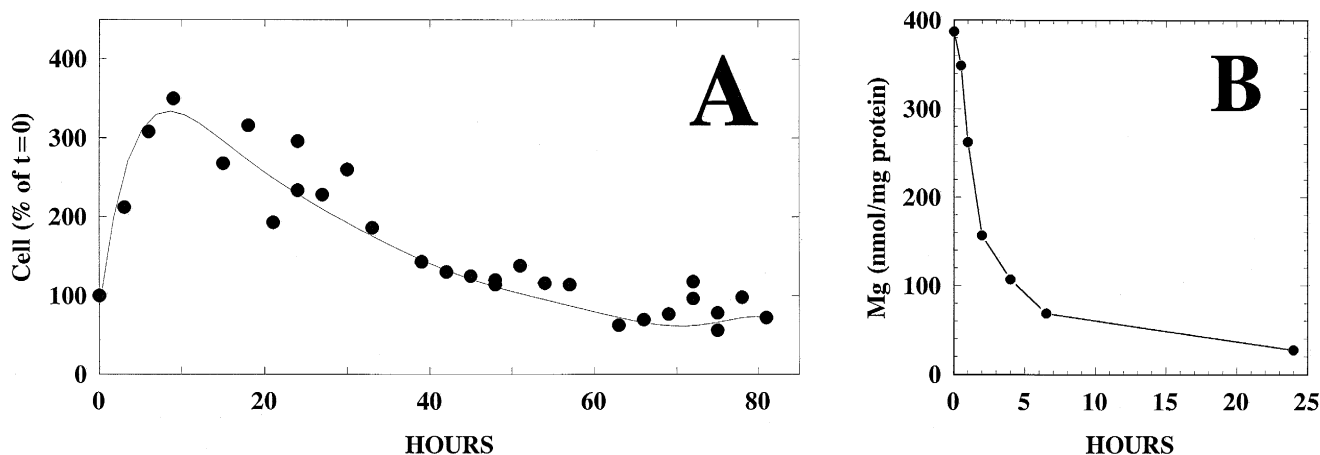


Fig. 4. Effect of Mg<sup>2+</sup>-free medium on cell viability and cellular Mg<sup>2+</sup>. (A) Cells grown in SD medium for several generations to an A<sub>600</sub> of 0.1–1.0 were shifted to Mg<sup>2+</sup>-free SD medium and at the indicated times aliquots of cells were plated onto YPD plates. After three days at 26°C, the number of viable cells (i.e., those that formed colonies) was determined. (B) The cellular Mg<sup>2+</sup> levels were determined at the indicated times after shifting to Mg<sup>2+</sup>-free SD media, using the dichlorophosphonazo III indicator as described in the legend to Fig. 2.

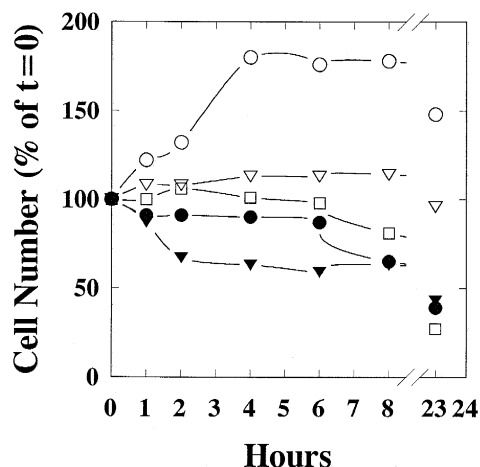


Fig. 5. Effect of preloading of cells with  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ , and polyphosphate on the ability of cells to grow in  $\text{Mg}^{2+}$ -free medium. Cells were grown in modified SD medium containing either 1 mM  $\text{MgSO}_4$ , 7 mM  $\text{KH}_2\text{PO}_4$  ( $\circ$ ); 20  $\mu\text{M}$   $\text{MgSO}_4$ , 7 mM  $\text{KH}_2\text{PO}_4$  (low  $\text{Mg}^{2+}$ ,  $\bullet$ ); 1 mM  $\text{MgSO}_4$ , 50  $\mu\text{M}$   $\text{KH}_2\text{PO}_4$  (low phosphate,  $\square$ ); 25  $\mu\text{M}$   $\text{MgSO}_4$ , 50  $\mu\text{M}$   $\text{KH}_2\text{PO}_4$ , (low  $\text{Mg}^{2+}$ , low phosphate,  $\nabla$ ); or 20  $\mu\text{M}$   $\text{MgSO}_4$ , 7 mM  $\text{KH}_2\text{PO}_4$ , 100 mM  $\text{CaCl}_2$  (high  $\text{Ca}^{2+}$ , low  $\text{Mg}^{2+}$ ,  $\blacktriangledown$ ). Cells were diluted into  $\text{Mg}^{2+}$ -free SD media and at the indicated times aliquots were removed and plated onto YPD plates to determine the number of viable cells.

in media containing 100 mM  $\text{CaCl}_2$  accumulate 200–300 nmol  $\text{Ca}^{2+}$ /mg in their vacuole [5], presumably displacing vacuolar  $\text{Mg}^{2+}$ .  $\text{Ca}^{2+}$ -loaded cells not only stop growth immediately upon transfer to  $\text{Mg}^{2+}$ -free medium, but display a 40% drop in viability within 4 h. These data demonstrate that conditions that reduce

$\text{Mg}^{2+}$  storage capacity by the vacuole decrease the cell's ability to continue to grow in  $\text{Mg}^{2+}$ -free medium.

The rate at which the cellular  $\text{Mg}^{2+}$  level is restored to about 400 nmol/mg after it is reduced to 200 nmol/mg is shown in Fig. 6.  $\text{Mg}^{2+}$  accumulation follows first order kinetics with a rate constant of  $1.4 \text{ h}^{-1}$  (Fig. 6B).

Polyphosphate accumulation by cells is dependent on the  $\text{Mg}^{2+}$  concentration of the growth medium [13–15]. The effect of  $\text{Mg}^{2+}$  concentration on polyphosphate accumulation is especially striking when cells grown in low phosphate are shifted to high phosphate [16]. Under these conditions there is a burst of polyphosphate synthesis ('overcompensation'), the amount of which is dependent on the  $\text{Mg}^{2+}$  concentration in the growth media. 'Over-compensation' polyphosphate synthesis is accompanied by increased  $\text{Mg}^{2+}$  accumulation (Fig. 7), demonstrating a relationship between vacuolar  $\text{Mg}^{2+}$  levels and polyphosphate levels. Polyphosphate synthesis during 'overcompensation' might be limited by the availability of  $\text{Mg}^{2+}$  to form a complex with polyphosphate.

Effect of  $\text{Mg}^{2+}$  on vacuolar and cellular  $\text{Ca}^{2+}$  accumulation Cellular  $\text{Ca}^{2+}$  accumulation is dependent on the  $\text{Mg}^{2+}$  concentration of the growth medium [17]. Vacuolar  $\text{Ca}^{2+}$  accumulation is mediated by the  $\text{Ca}^{2+}/2\text{H}^{+}$ -exchanger [5]. The  $K_m$  (25  $\mu\text{M}$ ) of the vacuolar  $\text{Ca}^{2+}$  transporter is much higher than the cytosolic  $\text{Ca}^{2+}$  concentration and depends on

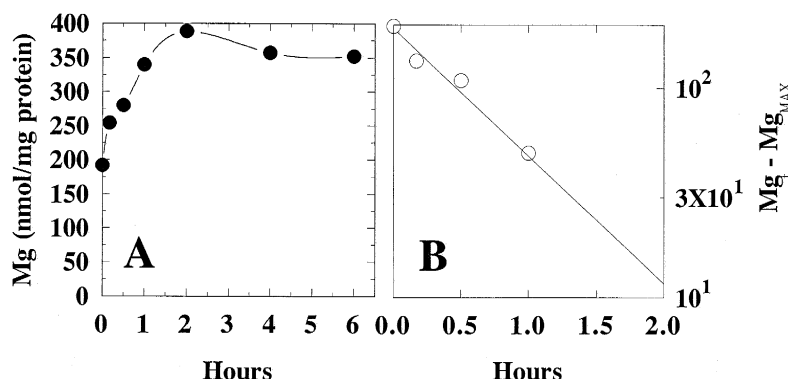


Fig. 6. Rate of  $\text{Mg}^{2+}$  accumulation by cells following a shift from  $\text{Mg}^{2+}$ -free medium to medium containing 1 mM  $\text{Mg}^{2+}$ . (A) Cells were grown in  $\text{Mg}^{2+}$ -free SD medium for several hours to reduce cellular  $\text{Mg}^{2+}$ , and then 1 mM  $\text{MgSO}_4$  was added. At the indicated times after addition of  $\text{Mg}^{2+}$ , aliquots of the cells were assayed for accumulated  $\text{Mg}^{2+}$  using the dichlorophosphonazo III assay as described in the legend to Fig. 2. (B) The rate of  $\text{Mg}^{2+}$  loading of the  $\text{Mg}^{2+}$ -deficient cells is measured by plotting the logarithm of the difference between the maximum  $\text{Mg}^{2+}$ -loading level ( $t = 2 \text{ h}$ ) and the level at the indicated time vs. the time.

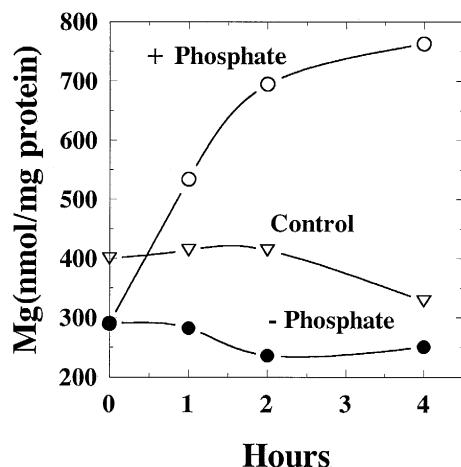


Fig. 7. Accumulation of  $\text{Mg}^{2+}$  by phosphate-starved cells following transfer into high phosphate medium. Cells were grown in SD medium + 10 mM  $\text{Mg}^{2+}$ , and a portion of the cells was then shifted to SD-phosphate + 10 mM  $\text{Mg}^{2+}$  for 4 h (○, ●). The control cells were left in the SD + 10 mM  $\text{Mg}^{2+}$  (▽). After the 4 h in phosphate-free medium, 7 mM phosphate was either added back (○) or the cells were maintained in the phosphate-free medium (●). The cellular  $\text{Mg}^{2+}$  levels were determined at the indicated times using the dichlorophosphonazo III assay as described in the legend to Fig. 2.

the  $\text{Mg}^{2+}$  concentration [5]; therefore, the rate of vacuolar  $\text{Ca}^{2+}$  uptake (Fig. 8A) depends on the  $\text{Mg}^{2+}$  concentration. As the free  $\text{Mg}^{2+}$  concentration increases from 0.2 mM to 9 mM the  $\text{Ca}^{2+}$  transport rate decreases at least 100-fold. The cellular  $\text{Ca}^{2+}$  loading level is also dependent on the  $\text{Mg}^{2+}$  concentration (Fig. 8B). When grown in YPD + 50 mM  $\text{Ca}^{2+}$  for 2 h, cells accumulate about 90 nmol  $\text{Ca}^{2+}$ /mg protein. As the  $\text{Mg}^{2+}$  concentration is increased from 0.6 mM to 5 mM, the accumulated  $\text{Ca}^{2+}$  decreases to a level too low to be detected with Arsenazo III (less than 2 nmol/mg) (Fig. 8B).

#### 4. Discussion

Based on the data reported here and from other laboratories, a model for the regulation of cytosolic  $\text{Mg}^{2+}$  in *S. cerevisiae* is proposed. Cytosolic  $\text{Mg}^{2+}$  is regulated by  $\text{Mg}^{2+}$  transport systems found in the vacuolar and plasma membranes. The free cytosolic  $\text{Mg}^{2+}$  concentration is unknown, but in general, eukaryotic cells are believed to have free cytosolic

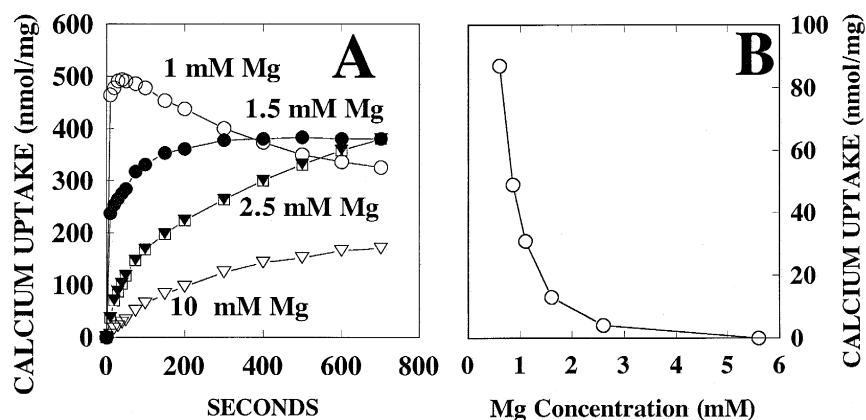


Fig. 8. Effect of  $\text{Mg}^{2+}$  concentration on the rate of vacuolar  $\text{Ca}^{2+}$  uptake and cellular  $\text{Ca}^{2+}$  accumulation. (A) Permeabilized partially regenerated spheroplasts were diluted 50-fold to a final protein concentration of 30  $\mu\text{g}/\text{ml}$  into 0.1 M KCl, 10 mM Hepes/Pipes/Mes (pH 7.0), 50  $\mu\text{M}$   $\text{CaSO}_4$ , 50  $\mu\text{M}$  Arsenazo III, 1 mM Na ATP and the indicated  $\text{Mg}^{2+}$  concentration at 27°C.  $\text{Ca}^{2+}$  uptake was monitored spectrophotometrically by measuring the decrease in the difference absorbance of the  $\text{Ca}^{2+}$ :Arsenazo III complex at 660 nm using 685 as reference wavelength. Baseline absorbance was established by addition of 5  $\mu\text{M}$  A23187, a  $\text{Ca}^{2+}$  ionophore that releases accumulated  $\text{Ca}^{2+}$ . The response of Arsenazo III to  $\text{Ca}^{2+}$  was calibrated by multiple additions of 2.5  $\mu\text{M}$   $\text{Ca}^{2+}$ . (B) Cells were grown to an  $A_{600}$  of 1.0 and diluted into YPD medium + 50 mM  $\text{CaCl}_2$  with  $\text{Mg}^{2+}$  added to the indicated concentration (YPD has about 0.6 mM  $\text{Mg}^{2+}$ ). After a 2-h incubation, cells were washed by centrifugation and resuspended to a density of  $10^7$  cells/ml in KCl solution containing 50  $\mu\text{M}$  Arsenazo III. Accumulated  $\text{Ca}^{2+}$  was released by adding 1 mg/ml digitonin to permeabilize all membranes and measured spectrophotometrically by monitoring the difference absorbance of the Arsenazo III- $\text{Ca}^{2+}$  complex at 660 nm/685 nm using an AMINCO-SLM dualbeam spectrophotometer.

$\text{Mg}^{2+}$  concentrations in the range of 0.1 to 1 mM. Since the cell volume is about 12  $\mu\text{l}/\text{mg}$  protein [10], the free  $\text{Mg}^{2+}$  would contribute 1.2–12 nmol  $\text{Mg}^{2+}/\text{mg}$  protein to the total cellular  $\text{Mg}^{2+}$  which, in cells grown in excess  $\text{Mg}^{2+}$ , is about 400 nmol/mg protein indicating only 0.3–3% of the cellular  $\text{Mg}^{2+}$  is free. In the cytosol,  $\text{Mg}^{2+}$  is bound mainly to ATP and to RNA. The  $\text{MgATP}$  concentration is about 1 mM [9] (12 nmol/mg protein). RNA binds about 0.2  $\text{Mg}^{2+}$  per phosphate residue under physiological conditions [18,19] and cells contain about 590 nmol nucleotide residues per mg protein [10], so that about 117 nmol  $\text{Mg}^{2+}/\text{mg}$  protein would be expected to be bound to RNA. The free  $\text{Mg}^{2+}$  and that bound to ATP or RNA would account for about 140 nmol/mg protein  $\text{Mg}^{2+}$ . When cells are transferred into  $\text{Mg}^{2+}$ -free medium, cell growth continues until the cellular  $\text{Mg}^{2+}$  level is reduced to about 80 nmol/mg, while cells growing in 10–30  $\mu\text{M}$   $\text{Mg}^{2+}$  have normal growth rates and a cellular  $\text{Mg}^{2+}$  content of about 90 nmol/mg. Cells grown in media containing 0.5 mM to 100 mM  $\text{Mg}^{2+}$  contain about 400 nmol/mg. About 75% of this  $\text{Mg}^{2+}$  appears to be complexed to polyphosphate in the vacuole. Cells can be loaded with 750 nmol  $\text{Mg}^{2+}/\text{mg}$  by inducing polyphosphate synthesis by transfer of phosphate-starved cells into media containing 7 mM phosphate.

The growth of cells varies with the  $\text{Mg}^{2+}$  concentration in the range of 1 to 30  $\mu\text{M}$ .  $\text{Mg}^{2+}$  transporters apparently can maintain 20- to 200-fold  $\text{Mg}^{2+}$  gradients across the plasma membrane. The polarization of the plasma membrane by the  $\text{H}^+$ -ATPase would provide a sufficient inside-negative membrane potential to form such gradients. Although the growth rate of cells does not significantly increase when the medium  $\text{Mg}^{2+}$  concentration increases from 0.1 to 1.0 mM, the cellular  $\text{Mg}^{2+}$  content increases 4-fold. The increased  $\text{Mg}^{2+}$  is likely due to increased polyphosphate-bound  $\text{Mg}^{2+}$  in the vacuole. The amount of vacuolar polyphosphate is about 0.1 mg/mg protein or about 1  $\mu\text{mol}$  phosphate residue per mg protein [5].  $\text{Mg}^{2+}$  bound to the polyphosphate provides a significant  $\text{Mg}^{2+}$  reservoir which can be mobilized when extracellular  $\text{Mg}^{2+}$  concentration is low. In  $\text{Mg}^{2+}$ -free medium, the cytosolic  $\text{Mg}^{2+}$  concentration is maintained by utilization of the vacuolar  $\text{Mg}^{2+}$ ; thus the vacuole appears to be important in the regulation of cytosolic  $\text{Mg}^{2+}$  under conditions of

low extracellular  $\text{Mg}^{2+}$ .  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  compete for the vacuolar storage sites but  $\text{Mg}^{2+}$  inhibits the  $\text{Ca}^{2+}/2\text{H}^+$  exchanger. Therefore, when the  $\text{Mg}^{2+}$  concentration is relatively high, vacuolar  $\text{Mg}^{2+}$ -loading apparently has priority over  $\text{Ca}^{2+}$ -loading unless the cytosolic  $[\text{Ca}^{2+}]$  is very high. At low  $\text{Mg}^{2+}$  concentrations,  $\text{Ca}^{2+}$  would displace  $\text{Mg}^{2+}$  thereby mobilizing the  $\text{Mg}^{2+}$  stores.

The plasma membrane  $\text{Mg}^{2+}$  transporter is regulated such that cells in  $\text{Mg}^{2+}$ -free medium do not leak  $\text{Mg}^{2+}$  out of the cell, and cells in media containing high  $\text{Mg}^{2+}$  concentrations do not over-accumulate  $\text{Mg}^{2+}$ .

Finally, it is interesting to note that when  $\text{Mg}^{2+}$  in the growth medium becomes limiting, the cells not only cease growing, but actually die. This is in contrast to starvation for most other nutrients where cells have mechanisms that allow them to enter  $G_0$  where they can survive for extended periods. Cells that enter  $G_0$  due to the lack of glucose remain viable in  $\text{Mg}^{2+}$ -free medium indicating that it is the dilution of cellular  $\text{Mg}^{2+}$  by growth that leads to cell death in  $\text{Mg}^{2+}$ -free medium.

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